To the Editor:—I read with considerable interest the article by Benavides et al. describing experiments of air bubble growth in water during exposure to 100% nitrous oxide, 100% xenon, or 50% xenon–50% oxygen. Although the experiments were nicely conducted, they explore a physics of gas flux in an unconstrained bubble permitted to grow spherically. Importantly, this geometry has limited biologic relevance for bubbles occluding vessels in the size range they have studied. The authors have referenced our previous work on xenon transport, but they have mistakenly interpreted the findings presented therein to indicate the growth of bubbles as spheres. Rather, that study presents some simulations for bubbles that are initially spherical and just fill the vessel lumen. Such bubbles cannot grow radially because they are constrained by the vessel wall and therefore elongate during growth while maintaining a fixed curvature on the interface. This results in a much different force balance across the gas-liquid interface and, hence, a different pressure condition on the interior of the bubble from that which occurs in the case of a time-varying interfacial shape, which the authors have studied. We have described these differences in our previous theoretical and experimental studies of intravascular gas embolism.

In addition, the initial internal gas content they have studied includes nitrogen, equilibrated with test solution A, but test solution B is nitrogen free. Hence, there are large gradients for nitrogen flux when the solutions are switched. We purposely avoided nitrogen as a component of either the bubble or perfusate in our predictions and considered only oxygen and xenon as the transportable species. Whereas others have studied growth of similarly unconstrained air bubbles during cardiopulmonary bypass, our work has not provided any data for direct comparisons such as the authors have made, based on the different gas constituents and the governing physics dictated by the shape constraint.

I find it fascinating, however, that they have couched their results in terms of bubble diameter growth. When transferred to the volume domain, one readily sees that the spherical bubbles exposed to 100% xenon or 100% nitrous oxide had grown to more than twice their initial volume in 25 min (figs. 2 and 3) and continued growing when the solutions were switched (downward arrow). The time required for this is surprisingly similar to the volume doubling times we reported for many of the cases we explored, despite the differences in our model and these experiments.

The curve fitting by a double exponential suggests that there will be continuous exponential growth of bubble diameter. So although the physics and gas transport are different from what we studied, the indication of the studies are the same.

David M. Eckmann, Ph.D., M.D., University of Pennsylvania, Philadelphia, Pennsylvania. eckmannmd@uphs.upenn.edu

References


In Reply:—Our article on the expansion of gas bubbles by xenon and nitrous oxide investigated how air bubbles of various dimensions in aqueous solution would expand when suddenly exposed to solutions containing certain gas mixtures (particularly mixtures containing xenon). The motivation behind this work was simple: Would air bubbles that were entrained while on cardiopulmonary bypass during cardiac surgery expand to a worrying extent if xenon were used during the procedure, hence potentially exacerbating damage caused by air emboli? Xenon has been proposed for use as a neuroprotectant, and the procedure, hence potentially exacerbating damage caused by air emboli. Xenon has been proposed for use as a neuroprotectant, and the procedure, hence potentially exacerbating damage caused by air emboli. Xenon has been proposed for use as a neuroprotectant, and the procedure, hence potentially exacerbating damage caused by air emboli. Xenon has been proposed for use as a neuroprotectant, and the procedure, hence potentially exacerbating damage caused by air emboli. Xenon has been proposed for use as a neuroprotectant, and the procedure, hence potentially exacerbating damage caused by air emboli. Xenon has been proposed for use as a neuroprotectant, and the procedure, hence potentially exacerbating damage caused by air emboli. Xenon has been proposed for use as a neuroprotectant, and the procedure, hence potentially exacerbating damage caused by air emboli. Xenon has been proposed for use as a neuroprotectant, and

Indeed, Dr. Eckmann and his colleagues have suggested exactly that, based on theoretical calculations that concluded that small gas bubbles would expand rapidly and indefinitely if they were trapped in fine blood vessels. (We fully understand that the model assumes that the bubbles are constrained by the size of the capillaries.) For example, their calculations suggest that a 50-nl bubble of oxygen exposed to 70% xenon–30% oxygen would grow to 250 nl in approximately 20 min with an ever-increasing rate of growth. Because we thought that these predictions were implausible, and because there were a large number of variables that had to be estimated, we conducted our experiments, which were designed to measure bubble growth directly under a well-defined set of conditions. We studied the expansion of both air and oxygen bubbles, and the results were similar; our data show bubble expansions of the order of 10% in diameter and 30% in volume under conditions likely to be encountered during cardiopulmonary bypass. We concluded that this is unlikely to represent a significant clinical problem.

We disagree with Dr. Eckmann’s claim that his calculations predict similar expansions to those we observed. Apart from the extent of the volume expansions that were predicted, their most striking aspect was the ever-increasing rates of expansion that seemed to predict unlimited bubble growth. In contrast, we observed limited bubble growth with volumes tending toward finite equilibrium values. Even making allowances for the differences between the model and the gas compositions, we believe our experimental observations probably better reflects reality than the theoretical calculations that Dr. Eckmann has published. Furthermore, in our recently published feasibility and tolerability clinical study involving exposure of cardiac surgical patients to xenon while on cardiopulmonary bypass, there was no

Drs. Franks and Maze have a financial interest in an Imperial College spin-out company (Protexon Ltd., London, United Kingdom) that is interested in developing clinical applications for medical gases, particularly xenon, and both are paid consultants for this activity. Drs. Franks and Maze also sit on a Strategic Advisory Board that advises Air Products, Allentown, Pennsylvania, on possible medical applications for gases, including anesthetic gases. The funding for the current study was provided by Carburos Metálicos, a wholly owned subsidiary of Air Products.
increase in the number of observed emboli, which would have occurred had the bubble size of gas emboli increased significantly.5

Nicholas P. Franks, Ph.D., F.Med.Sci.,* Rodrigo Benavides, M.D., Mervyn Maze, F.R.C.P., F.R.C.A., F.Med.Sci. Imperial College London, London, United Kingdom. n.franks@ic.ac.uk

References


(Accepted for publication June 19, 2006.)
 ally require longer time for placement compared with the left-sided DLT.3,4
The results of the study4 are rather concerning, because training and expo-
sure for residents largely consist of the use of the DLT, which is used more
often and has been in clinical practice for a longer period of time. Because
a combination of unfamiliarity with tracheobronchial anatomy and skill in
fiberoptic bronchoscopy was responsible for most of the malpositions,1
perhaps training for residents should concentrate on building a stronger
foundation in basic knowledge and skill. Personal experience with the routine
use of video-bronchoscopy for confirmation of DLT position has shown that
it greatly facilitates the learning process. Anatomy can be demonstrated
clearly, and due to simultaneous viewing by the attending and resident, there
is better appreciation of the steps involved in the identification and correction
of any malposition problems. The use of a virtual bronchoscopy simulator
may also be useful in skill acquisition in flexible bronchoscopy.2

Ju-Mei Ng, F.A.N.Z.C.A., Singapore General Hospital, Singapore.
ng.ju.mei@sgh.com.sg

In Reply—We thank Drs. Ng and Neustein for their interest in our research.3 Addressing Dr. Ng’s letter, we fully agree with his comments that
more attention should be given while training residents or staff
anesthesiologists in lung isolation techniques, with particular emphasis on
tracheobronchial anatomy and skills in fiberoptic bronchoscopy. In fact,
after we published our study,4 we designed a second study based on the
lessons that we learned previously. Because the previous study did not
demonstrate any advantage between the left-sided double-lumen endo-
tracheal tubes and bronchial blockers (Univent® [Vitaid Ltd., Lewiston, NY]
and Arndt® [Cook Critical Care, Bloomington, IN] blocker) and because
the most common device used for lung isolation is the left-sided double-
lumen endotracheal tube, we are currently conducting a new study,
involving the use of left-sided double-lumen endotracheal tubes among
anesthesiologists with limited experience in thoracic anesthesia (i.e., less
than two lung isolation device cases per month). In this randomized study,
one group of anesthesiologists has been assigned to a tutorial in the
simulator facility at The University of Iowa (Iowa City, Iowa), providing a
tutorial demonstration and hands-on practice in proper placement of a
double-lumen endotracheal tube with the aid of flexible fiberoptic bron-
choscopy techniques on a mannequin model. The second group has been
assigned to self-training using a DVD that was made by one of the authors
(J.H.C.) that includes placement of left-sided double-lumen endotracheal
tubes along with a detailed description of fiberoptic bronchoscopy tech-
niques. It is our hope to have a definitive answer to determine which
method (simulator training vs. DVD self-training) facilitates placement of
double-lumen endotracheal tubes for anesthesiologists with limited tho-
racic anesthesia experience.

Our study has shown that one of the limitations of anesthesiologists
with limited experience in thoracic anesthesia is unfamiliarity with
bronchial anatomy.3 In principle, every anesthesiologist resident or
staff member should know the anatomical distances pertaining to the
airway; for example, in an average subject, the distance from the
incisors to the vocal cords is 15 cm, and the distance from the vocal
cords to the tracheal carina is approximately 12 cm. The distance from
the tracheal carina to the takeoff of the right upper lobe bronchus is an
average of 1.5 cm in females and 2 cm in males. The distances from the
carina to the takeoff of the left upper and lower bronchi are an
average of 4.5 and 5.0 cm, respectively.

Furthermore, when looking through the fiberoptic bronchoscope,
the only early structure in the right mainstem bronchus that has three
orifices is the right upper lobe bronchus: These are the apical, the
anterior and the posterior bronchi. If every anesthesiologist recognized
this anatomical structure, fewer problems would be encountered
when inserting lung isolation devices. Adding a shared demonstration

References

used by anesthesiologists with limited thoracic experience: Comparison of dou-
ble-lumen endotracheal tube, Univent® torque control blocker, and Arndt wire-
guided endobronchial blocker. Anesthesiology 2006; 104:261–6
double-lumen tube for one-lung ventilation in a variety of noncardiac thoracic
blocker compared to double-lumen tube for one-lung ventilation during thora-
the torque control blockor Univent and the wire-guided blocker. Anesth Analg
2003; 96:283–9

(Accepted for publication June 22, 2006.)
number of malpositions was quite low for both tube types and did not differ between the groups. Therefore, the choice of device for lung isolation in the right mainstem bronchus does not matter when a cardiothoracic anesthesiologist places these devices. This concept might not apply to anesthesiologists with limited experience in thoracic anesthesia, but to our knowledge, this has not been tested scientifically.

Javier H. Campos, M.D.,* Ezra A. Hallam, B.A., Timothy Van Natta, M.D., Kemp H. Kernstine, M.D., Ph.D. 'The University of Iowa Roy J. and Lucille A. Carver College of Medicine, Iowa City, Iowa. javier-campos@uiowa.edu

References


(Accepted for publication June 22, 2006.)

---

**Ambesh Maneuver during Subclavian Vein Catheterization Successfully Prevents and Detects Misplacement of the Catheter into Ipsilateral Internal Jugular Vein**

To the Editor—During subclavian vein catheterization, one of the most common misplacements of the catheter is into the ipsilateral internal jugular vein (IJV). Chest radiography is performed to identify the exact location of the catheter and procedure-related complications. Misplaced catheters have increased risks of thrombophlebitis in addition to impairment of the central venous pressure (CVP) measurement. Recently, Ambesh et al. showed that manual occlusion of the ipsilateral IJV in the supraclavicular fossa during and after insertion of subclavian vein catheter is successful in preventing and diagnosing the misplacement of the subclavian vein catheter into the IJV, respectively. Since Ambesh et al. developed this maneuver and reported excellent results, no other study has validated these results. Therefore, in a randomized and controlled study, we tested whether the Ambesh maneuver is successful in preventing and diagnosing the misplacement of a subclavian vein catheter into the IJV.

After approval by the Ethics Committee (King George's Medical University, Lucknow, India), 300 adult patients of either sex scheduled to undergo central venous cannulation through the subclavian approach were randomly allocated into two groups of 150 each. Informed verbal consent was obtained from all patients before the procedure. In a patient lying supine with a 15°–20° Trendelenburg position and the head turned to the left, the junction of the mediasternal one third and lateral two thirds of the clavicle in the right infraclavicular area was chosen as the puncture point. A 18-gauge introducer needle was inserted at this point and directed toward the sternooclavicular joint. After free flow of venous blood, the J-tip guide wire was threaded through the cannula into the subclavian vein. In the Ambesh maneuver group of patients, the ipsilateral IJV was occluded, as described by Ambesh et al. during threading of the J guide wire, whereas in control group, no such maneuver was performed. The subclavian vein catheter was then railroaded over the guide wire. The catheter was then connected with a transducer, and the CVP value and waveform pattern were observed. Next, the Ambesh maneuver was reapplied for approximately 10 s, and changes in CVP value and waveform pattern were noted. If there was an increase in CVP value by more than 3 cm H₂O along with flattening of the waveform, it was presumed that the catheter tip was misplaced into the ipsilateral IJV. At the end of the procedure, chest radiography was performed, and position of the catheter was identified in all patients. The characteristics of the patients were analyzed using the Student t-test and Fisher exact test. P values were two-tailed, and \( P > 0.05 \) was considered significant.

The age, sex, and body weight of the patients were comparable in two groups. Five patients in the control group and 7 patients in the Ambesh maneuver group could not be cannulated; therefore, 145 patients in the control group and 143 patients in the Ambesh maneuver group were analyzed. Chest x-ray films showed that in control group, there were 10 (6.9%) misplaced catheters, 9 (6.2%) in ipsilateral IJV and 1 (0.7%) in opposite subclavian vein, whereas in the Ambesh maneuver group, there were 2 (1.4%) (95% confidence interval, 1.4–6.9%; \( P < 0.05 \)) misplaced catheters, both in the opposite subclavian vein and none in the IJV. The operator experienced difficulty in inserting the guide wire in 3 patients of control group and 9 patients of Ambesh maneuver group. The withdrawal and reinsertion of the guide wire and catheter were easy.

The correct placement of the central venous catheter is an essential prerequisite for accurate monitoring of CVP and long-term use of the catheter. Misplacement of the tip may enhance the risk of clot formation, thrombophlebitis, and catheter erosion in addition to impaired CVP measurement. Recently, Domino et al. reported that the proportion of malpractice claims related to central catheters and vascular access injury has increased significantly. The incidence of malpositioning of CVP catheters through the infraclavicular technique of the subclavian vein varies between 4% and 8%. Our study shows a 6.9% incidence of misplacement of subclavian vein catheter through the right infraclavicular approach, and most of the misplacements were in the ipsilateral IJV (6.2%). The operator encountered difficulty in insertion of the guide wire in only 3 patients without IJV occlusion and in 9 patients with IJV occlusion. It becomes obvious that the guide wire in some of these 9 patients of the IJV occlusion group was intending to go into the ipsilateral IJV, but the IJV was occluded manually. The occlusion of ipsilateral IJV in the supraclavicular area may have prevented the cephalad insertion of the guide wire and therefore the subclavian vein catheter into the IJV.

We conclude that the Ambesh maneuver is a simple, inexpensive, and handy bedside technique that helps in preventing and diagnosing the misplacement of a subclavian vein catheter into the IJV. We strongly believe that the Ambesh maneuver should be used in all patients undergoing subclavian vein catheterization.

Dinesh K. Singh, M.D.,* Monica K. Kohli, M.D., Vasudha Singhal, M.D., Vineeta Singh, M.D., Alka Rani, M.D. ‘King George’s Medical University, Lucknow, India. dksingh_kgmu@rediffmail.com

References


---

Support was provided solely from institutional and/or departmental sources.
To the Editor—American inventor and businessman Thomas Alva Edison is legendary for his contributions to such technologies as the lightbulb, the telephone, the phonograph, and motion pictures, among many others. In his lifetime, Edison obtained 1,093 US patents and some 1,239 patents in other countries. Little known among these efforts was his “improved anesthetic compound.”

In the summer of 1882, George F. Shrady (Founder and Editor, Medical Record 1866–1904) (1837–1907), reported that Thomas Edison invented a new anesthetic made of chloroform, ether, alcohol, and camphor and had applied for British and German patents. The witty but misinformed editor added, “Edison may wish to use it on his stockholders until electric light was in successful operation.”

In fact, the “anesthetic” actually was an analgesic liniment that Edison had prepared in early 1878. He named it Polyform and advertised it for “neurologic pain.” Polyform was a mixture of chloroform, ether, camphor gum, alcohol, chloral hydrate, morphine, and oils of peppermint and clove. Edison believed that his compound’s various analgesics would potentiate each other and that the mixture would attack pain in a “shotgun manner.”

In 1879, Edison applied for a US patent but, for unknown reasons, withdrew his application shortly thereafter. In February 1880, the British patent No. 599 was granted to his London agents for a slightly modified compound. The editor of the Medical Record was misinformed: Edison did not apply for a German patent (written personal communication, Hubert Rothe, Director, Information Department, German Patent and Trademark Office, Munich, Germany, May 2004).

Topical ether and, especially, chloroform had been widely used for musculoskeletal and neurologic pains since their discovery. At the time of Edison’s invention, not only were liniments of chloroform and of camphor used in the United States, but there also existed lotions made of chloroform, camphor, ether, alcohol, morphine, and chloral hydrate. One, Sankt Jakob Oel, had been popular in Germany since the mid-1870s. It was marketed in the United States during the 1880s under the name of St. Jacob’s Oil by the firms of C.A. Voegeler in Baltimore and Kroeger Ltd. of Cincinnati. Its formula is given in several US formularies. Whether Edison knew of St. Jacob’s Oil when he invented his Polyform or whether he learned of its existence later on is unknown. The latter may explain why he did not apply for a German patent and withdrew his US application.

Ray J. Defalque, M.D., M.S., Amos J. Wright, M.L.S.,* School of Medicine, University of Alabama at Birmingham, Birmingham, Alabama. ajwright@uab.edu

The authors thank Gregory J. Higby, Ph.D. (Professor and Director of the American Institute of the History of Pharmacy, Madison, Wisconsin), Michael A. Flannery, Ph.D. (Associate Professor and Associate Director for Historical Collections, University of Alabama at Birmingham, Birmingham, Alabama), and Hubert Rothe (Director, Information Department, German Patent and Trademark Office, Munich, Germany) for their valuable information.

References

2. Edison’s anaesthetic. Edited by Shrady GF. Medical Record 1882; 22:474

*School of Medicine, University of Alabama at Birmingham, Birmingham, Alabama.
We report a case of acute upper gastrointestinal bleeding from mucosal tears resulting from gastric overdistention after inadvertent esophageal intubation. To our knowledge, this complication has not been previously described and should be included in the differential diagnosis of upper gastrointestinal bleeding in critically ill patients who have had inadvertent esophageal intubation or cardiopulmonary resuscitation. The combination of history of esophageal intubation, involvement of the lesser curvature, and similarity to the mucosal tears seen by Green and Tendler in during percutaneous endoscopic gastrostomy tube placement support this diagnosis. In addition, in cases of upper gastrointestinal bleeding when a history of inadvertent esophageal intubation is obtained, gastric perforation should be ruled out with abdominal imaging before performing upper endoscopy.

Daniel K. Mullady, M.D.,* James B. McGee, M.D. “University of Pittsburgh School of Medicine, Pittsburgh, Pennsylvania. mulladylk@upmc.edu

References
4. Gilbert TB: Gastric rupture after inadvertent esophageal intubation with a jet ventilation catheter. Anesthesiology 1998; 88:537–8

Fig. 1. Retroflexed view revealing mucosal tears along the lesser curvature of the stomach caused by overdistention after inadvertent esophageal intubation.

Anesthesia in a Patient with Gitelman Syndrome

To the Editor.—We would like to report our recent experience with a patient with a rare disorder known as Gitelman syndrome. Gitelman syndrome, a variant of Bartter syndrome, is a congenital autosomal recessive disorder characterized by hypokalemia, hypomagnesemia, and hypocrenina associated with metabolic alkalosis. Unlike Bartter syndrome, which presents in the neonatal period and childhood up to 5 yr of age, Gitelman syndrome presents in early adulthood. The two syndromes may also be distinguished from each other because Gitelman syndrome presents with hypomagnesemia and hypocreninuria, whereas Bartter syndrome presents with normal serum magnesium and high urinary calcium. Patients with Gitelman syndrome usually present with cramps, fatigue, muscle weakness, and carpopedal spasms.

A 47-yr-old woman recently presented to us for repair of nasoacral duct stenosis during general anesthesia. Her medical history was significant for diabetes, gastroesophageal reflux disease, and thyroid disease. Her surgical history included a hysterectomy, an appendectomy, a thyroidectomy, a breast biopsy, and an exploratory laparotomy for ovarian cancer. She described her symptoms as cramps in her legs. She related that 2 yr ago during the surgery for her thyroid, she had a cardiac arrest. The patient stated that her cardiac arrest had occurred because of unrecognized hypokalemia and hypomagnesemia.

Her medications included potassium chloride, magnesium, spironolactone, levothyroxine, glyburide, and lensoprazole. Her vital signs included a blood pressure of 117/77 mmHg, pulse of 98 beats/min, respiration of 18 breaths/min, and temperature of 97.9°F. Her laboratory studies showed the following: white blood cells, 10.6 (4.1–11.2); hemoglobin, 15.3 g/dl (11.5–15.1); hematocrit, 45.0% (35–46); platelets, 503 ×103 (400–900); Na, 137 mEq/l (136–145); K, 4.2 mEq/l (3.5–5.1); Cl, 96 mEq/l (98–107); HCO3–, 27 (20–27); blood urea nitrogen, 20 mg/dl (6–20); creatinine, 0.8 (0.5–1.2); glucose, 68 mg/dl (65–115); Ca, 10.9 mEq/l (8.8–10.5); Mg, 1.9 mEq/l (1.3–2.1); and urine creatinine, 181 mEq/l. She was 4 ft 11 in tall and weighed 70 kg.

In the operating room after placement of an electrocardiograph, noninvasive blood pressure cuff, and pulse oximeter, general anesthesia was induced using propofol and fentanyl with rocuronium for muscle relaxation and for tracheal intubation. Desflurane was used...
for maintenance of anesthesia. At the end of the case, the muscle relaxant was reversed using glycopyrrolate and neostigmine. When the patient was awake and responding to commands, her trachea was extubated in the operating room without any complications.

Although the patient’s current magnesium level was only slightly decreased, we decided not to replace it. Her current potassium level was also normal. She was aware of her diagnosis of Gitelman syndrome and was taking potassium and magnesium replacement. Review of the literature reveals that ventricular tachycardia has occurred in patients with Gitelman syndrome when potassium and magnesium levels are low. This potentially fatal arrhythmia must be recognized and treated early.

References

2. Naessens M, Steels P, Verbereckmoes R, Vanreinterghem Y, Kuypers D: Bar-

(Accepted for publication June 28, 2006.)