

T and the $g_n(\bar{v})$ are differentiable and also integrable in closed form, subject to the restriction that

$$p \rightarrow RT/\bar{v} \text{ as } \bar{v} \rightarrow \infty. \quad (38)$$

All known analytic empirical thermal equations of state satisfy these conditions. From equation (37) one obtains

$$\left(\frac{\partial s}{\partial \bar{v}}\right)_T = \left(\frac{\partial p}{\partial T}\right)_{\bar{v}} = \sum_n f'_n(T)g_n(\bar{v}), \quad (39)$$

and

$$\left(\frac{\partial p}{\partial \bar{v}}\right)_T = \sum_n f_n(T)g'_n(\bar{v}). \quad (40)$$

The caloric equation of state is obtained in terms of the zero pressure specific heat at constant volume $c_{\bar{v}}^+(T)$, where $c_{\bar{v}}^+(T)$ and $c_{\bar{v}}^+(T)/T$ are both integrable in closed form. The caloric equation of state can then be written as

$$u = u^+(T) + \sum_n [Tf'_n(T) - f_n(T)] \int_{\infty}^{\bar{v}} g_n(\bar{v}) d\bar{v}, \quad (41)$$

where the zero pressure internal energy $u^+(T)$ is given by

$$u^+(T) = \int_0^T c_{\bar{v}}^+(T) dT. \quad (42)$$

The corresponding expression for $c_{\bar{v}}$ is

$$c_{\bar{v}} = c_{\bar{v}}^+(T) + T \sum_n f''_n(T) \int_{\infty}^{\bar{v}} g_n(\bar{v}) d\bar{v}. \quad (43)$$

The speed of sound c can then be determined from

$$c^2 = -\bar{v}^2 \left[\left(\frac{\partial p}{\partial \bar{v}}\right)_T - \frac{T}{c_{\bar{v}}} \left(\frac{\partial p}{\partial T}\right)_{\bar{v}}^2 \right]. \quad (44)$$

The specific heat at constant pressure c_p is most simply calculated from

$$c_p = -\frac{c_{\bar{v}}c^2}{\bar{v}^2 \left(\frac{\partial p}{\partial \bar{v}}\right)_T}. \quad (45)$$

The entropy can be obtained by integrating equation (38). It can be expressed as

$$s = s(T_0, \bar{v}_0) + s^+(T) - s^+(T_0)$$

$$+ \sum_n \left[f'_n(T) \int_{\infty}^{\bar{v}} g_n(\bar{v}) d\bar{v} - f'_n(T_0) \int_{\infty}^{\bar{v}_0} g_n(\bar{v}) d\bar{v} \right], \quad (46)$$

where $s^+(T)$ is given by the indefinite integral

$$s^+(T) = \int \frac{c_{\bar{v}}^+(T)}{T} dT, \quad (47)$$

and T_0, \bar{v}_0 is some arbitrary reference state.

For isentropic flow from a reservoir with T_0 and p_0 given, the corresponding value of \bar{v}_0 would in general be obtained from equations (37) and (40) using Newton's iteration. Instead of integrating

$$\left(\frac{\partial T}{\partial \bar{v}}\right)_s = -\frac{T}{c_{\bar{v}}} \left(\frac{\partial p}{\partial T}\right)_{\bar{v}} \quad (48)$$

to obtain $T(\bar{v})$, as is done in the discussed paper, it is much simpler to solve for $T(\bar{v})$ (or $\bar{v}(T)$) using Newton's iteration from equation (46) by setting

$$s - s(T_0, \bar{v}_0) = 0. \quad (49)$$

The choice of \bar{v} or T as independent variable depends on the nature of equation (37) and the function $c_{\bar{v}}^+(T)$. For the particular equations of Bober and Chow, the equation relating T and \bar{v} can be written as

$$s^+(T) - s^+(T_0) + R \ln \left(\frac{\bar{v} - b}{\bar{v}_0 - b} \right) + \frac{a}{2b} \left[T_0^{-3/2} \ln \left(1 + \frac{b}{\bar{v}_0} \right) - T^{-3/2} \ln \left(1 + \frac{b}{\bar{v}} \right) \right] = 0, \quad (50)$$

where

$$s^+(T) = R \left[(A_0 - 1) \ln T + A_1 T + \frac{1}{2} A_2 T^2 + \frac{1}{3} A_3 T^3 + \frac{1}{4} A_4 T^4 \right]. \quad (51)$$

For this case it would appear to be more efficient to choose T as the independent variable.

Acknowledgment

This work was supported by NASA Ames Research Center under Contract NAS 2-11555.

Authors' Closure

The authors would like to thank Dr. Vinokur for his eloquent generalization of the thermodynamics involved in the problem of a "Nonideal Isentropic Gas Flow Through Converging-Diverging Nozzles." In fact, as a result of his review, we were able to simplify our original version of the paper. In a more recent paper, one entitled, "Nonideal Gas Effects for the Venturi Meter," our approach follows more closely the method that is outlined in his Comment.