

## Measurement of H<sub>2</sub> consumption and its role in continuous fermentative hydrogen production

J. T. Kraemer and D. M. Bagley

### ABSTRACT

To maximise the yield from fermentative H<sub>2</sub> production, H<sub>2</sub> consumption must be minimised. This work demonstrated for the first time that H<sub>2</sub> consumption exists in an established continuous-flow biohydrogen system. The rate of H<sub>2</sub> consumption was found to be related to the concentration of CO<sub>2</sub>, with H<sub>2</sub> consumption inhibited at both low and high CO<sub>2</sub>. N<sub>2</sub> sparging of the continuous reactor at 31 mL/min/L-liquid increased the H<sub>2</sub> yield from 1.31 to 1.87 mol H<sub>2</sub>/mol glucose, but did not significantly change the *in-situ* rate of H<sub>2</sub> consumption (0.07–0.09 mM/h). Assuming sparging completely inhibited H<sub>2</sub> consumption, it could only account for 2–11% of the H<sub>2</sub> yield increase during sparging, based on H<sub>2</sub> consumption rates measured in the reactor and in vials. Therefore, H<sub>2</sub> consumption may be of minor concern for continuous biohydrogen systems.

**Key words** | acetogenesis, activity test, carbon dioxide, fermentation, hydrogen, sparging

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### INTRODUCTION

Renewable energy sources are currently receiving a lot of interest because of global climate change. Anaerobic digestion is a natural microbial degradation process which can convert organic wastes into renewable energy in the form of hydrogen and methane gases. H<sub>2</sub> is appealing because it does not release CO<sub>2</sub> during its utilization (although CO<sub>2</sub> is released during its formation).

H<sub>2</sub> is produced during the fermentation step of anaerobic digestion. Considerable research has been devoted to fermentative H<sub>2</sub> production recently and the current state of knowledge has been reviewed (Hawkes *et al.* 2007; Kraemer & Bagley 2007). To be successful, these systems must promote conditions conducive to H<sub>2</sub> production while minimising H<sub>2</sub>-consumption. Methanogenesis is the most prevalent route of H<sub>2</sub> consumption in anaerobic systems. Methanogens can be killed by heat treatment of the inoculum, but current evidence indicates this practice may not maximise the H<sub>2</sub> yield (Kraemer & Bagley 2007). Instead, operation of a continuous reactor at low pH (~5.5) and an appropriate retention time (~6–12 h) can wash out the methanogens (Kraemer &

Bagley 2007). However, there are other routes of H<sub>2</sub> consumption in anaerobic systems and these must be investigated in order to maximise H<sub>2</sub> yields.

Homoacetogenesis is another H<sub>2</sub>-consuming pathway that was observed in a batch biohydrogen system by Oh *et al.* (2003) for both heat-treated and non-heat-treated inocula. This work was followed up by Park *et al.* (2005) who observed that acetogenesis and concurrent H<sub>2</sub> removal in a heat-treated batch culture was prevented when the headspace CO<sub>2</sub> was removed using a KOH trap. These observations have led to the proposition that N<sub>2</sub> sparging in a continuous-flow system might decrease the dissolved concentrations of H<sub>2</sub> and CO<sub>2</sub> sufficiently to decrease the substrate available to acetogens (Hussy *et al.* 2003; Park *et al.* 2005; Kim *et al.* 2006; Kraemer & Bagley 2006).

Although H<sub>2</sub> consumption has been observed in batch culture, what is still unknown is whether H<sub>2</sub> consumption exists in a continuous-flow biohydrogen system. If H<sub>2</sub> consumption does exist, what is its rate and to what extent does it decrease the H<sub>2</sub> yield? And does N<sub>2</sub> sparging decrease H<sub>2</sub> consumption as proposed in the literature?

The objective of this work was to answer these fundamental questions.

## METHODS

### Reactor operation

A completely-mixed reactor (2 L liquid) was inoculated with methanogenic sludge (no heat treatment) and operated at 25°C, 10 h hydraulic retention time, pH 5.5, and mineral oil as antifoamant. The details of the reactor have been previously described (Kraemer & Bagley 2006). The substrate was glucose at 10 g/L supplemented with the nutrient solution of Kraemer & Bagley (2006). Nitrogen gas (99.999% pure, Praxair) was sparged into the reactor using a fritted-cylinder dispersion tube (#39533-12C, Corning Inc.). The off-gas was measured continuously using a FLOCELL<sup>®</sup> bubble meter (Challenge Environmental Systems, Inc.) without sparging or a differential pressure flowmeter (#32446-27, Cole-Parmer Instrument Co.) during sparging. The reactor was operated without sparging for 23 days before the work described here.

Two experimental runs of 14 days each were performed, with the first run using N<sub>2</sub> sparging at 31 mL/min/L-liquid while the second run did not use sparging. Analytical measurements were taken about every other day starting on the 3rd day of each run. Activity tests (described below) were performed during each run.

### Analytical measurements

Headspace and dissolved gas concentrations were measured according to Kraemer & Bagley (2006). Gas samples were taken using a SampleLock syringe (Hamilton Co.). Total and volatile suspended solids (TSS, VSS) were measured according to *Standard Methods for the Examination of Water and Wastewater* (1998).

Glucose was analyzed enzymatically using 25 µL of centrifuged sludge supernatant (20,000 g, 4°C, 20 min) and 4.0 mL glucose hexokinase reagent (Infinity, Thermo Electron Corp.) dispensed at 4°C. The liquid was vortexed for 1 s, incubated at 37°C for 3 min, and absorbance measured at 340 nm.

Acetone, alcohols (ethanol, 1-propanol, 1-butanol), and volatile fatty acids (acetic, propionic, *i*-butyric, *n*-butyric, *i*-valeric, *n*-valeric) were measured by static headspace-gas chromatography according to Cruwys *et al.* (2002). Modifications were made to accommodate analysis of acetone and alcohols, including the use of 2-methyl-1-butanol as alcohol internal standard (100 µL at 2,000 mg/L) and using a temperature program of 45°C for 1.8 min, 45°C/min to 140°C, and 10°C/min to 166°C. Other differences were: needle temperature 95°C, Zebron FFAP column (Phenomenex), one distilled water wash blank between samples, and use of ordinary least-squares regression.

### Measurement of H<sub>2</sub> consumption

H<sub>2</sub>-consuming activity was measured either using 22 mL headspace vials (“in-vial”) or directly in the reactor (“in-reactor”). For “in-vial” tests, the vials were capped with butyl rubber septa and aluminum crimp caps (Supelco) and purged with a 90% N<sub>2</sub>/10% CO<sub>2</sub> gas mixture (Praxair). Then 10 mL of reactor sludge, obtained with a gastight syringe, was injected while venting the vial headspace to maintain ~ 1 atm pressure. After the addition of 0.5 mL H<sub>2</sub> gas (99.999%, Praxair) and 0.5 mL CH<sub>4</sub> gas as internal standard (99.97%, Praxair), the vials were kept horizontal and mixed at 125 rpm in an orbital shaker (New Brunswick Scientific) at 20°C. The vial headspace gas compositions were measured approximately every 30 min either immediately or after 16–18 h depending on the purpose of the test. For some tests, 2nd and 3rd feedings of 0.3–0.5 mL H<sub>2</sub> were performed and the headspace measured 4–5 times after each feeding. For “in-reactor” H<sub>2</sub> consumption tests, the liquid flows and N<sub>2</sub> sparging (if in use) were stopped and dissolved and headspace H<sub>2</sub> and CO<sub>2</sub> concentrations were measured every 15–20 min.

## RESULTS & DISCUSSION

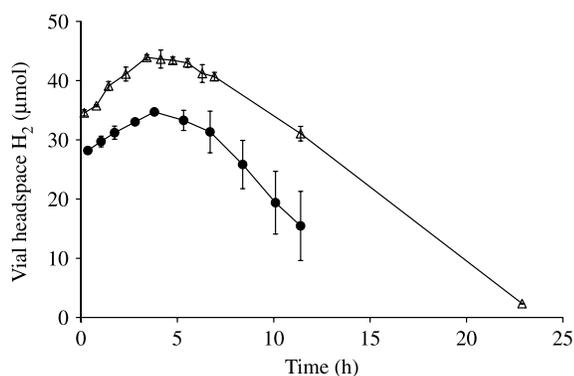
The performance of the continuous-flow biohydrogen reactor with and without sparging is summarized in Table 1. Sparging significantly increased the H<sub>2</sub> yield, decreased the dissolved and headspace H<sub>2</sub> and CO<sub>2</sub> concentrations, and increased the amount of acetate produced. There was no significant change in the *n*-butyrate

**Table 1** | Performance of the fermentative hydrogen production bioreactor with and without N<sub>2</sub> sparging at 31 mL/min/L-liquid. Values are average  $\pm$  95% confidence intervals (CI)

	No Sparging	With Sparging
H <sub>2</sub> yield, mol H <sub>2</sub> /mol glucose converted	1.31 $\pm$ 0.04	1.87 $\pm$ 0.05
Headspace gas composition, mol%		
H <sub>2</sub>	58.8 $\pm$ 0.7	9.7 $\pm$ 0.5
CO <sub>2</sub>	41.2 $\pm$ 0.7	7.2 $\pm$ 0.3
Dissolved gas concentration, mM		
H <sub>2</sub>	0.72 $\pm$ 0.04	0.36 $\pm$ 0.03
CO <sub>2</sub>	5.5 $\pm$ 0.2	0.75 $\pm$ 0.06
VSS, mg/L	125 $\pm$ 46	47 $\pm$ 5
Soluble Constituents, mg/L		
Acetate	1,143 $\pm$ 60	1,327 $\pm$ 67
n-butyrate	2,235 $\pm$ 131	2,322 $\pm$ 221

concentration with sparging. Regardless of sparging, the glucose conversion efficiency was 100% and the other compounds measured were <90 mg/L.

Figure 1 shows that sludge taken from an established continuous-flow fermentative hydrogen production bioreactor and placed in an activity test vial will consume H<sub>2</sub>. Initially, the headspace H<sub>2</sub> concentration increased, which can be attributed to continued H<sub>2</sub> production by the sludge and mass transfer of H<sub>2</sub> out of solution because the initial vial headspace H<sub>2</sub> concentration (2.4 nmol/ $\mu$ L) was lower than that in the liquid (~3.5 nmol/ $\mu$ L with sparging, ~22 nmol/ $\mu$ L without sparging). H<sub>2</sub> consumption was



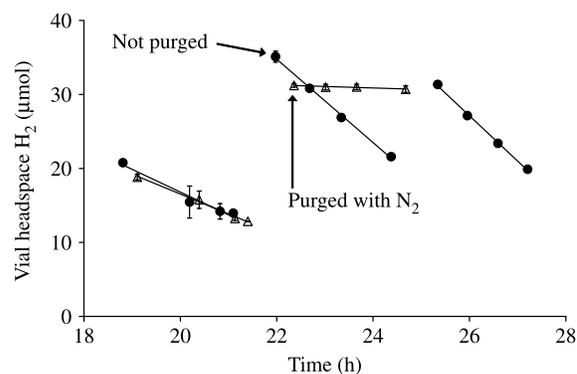
**Figure 1** | In-vial H<sub>2</sub> consumption by sludge taken from a continuous fermentative H<sub>2</sub> bioreactor that was sparged with N<sub>2</sub> at 31 mL/min/L ( $\Delta$ ) or un-sparged ( $\bullet$ ). Error bars show the range.

observable after 4 hours, and this lag time appeared to be similar regardless of whether the sludge came from a sparged or un-sparged reactor.

In another test, repeated additions of H<sub>2</sub> to the activity test vials were also consumed (Figure 2 – closed circles). Furthermore, the rate at which H<sub>2</sub> was consumed increased with each feeding (i.e. increased with time). In Figure 2, the rates of H<sub>2</sub> consumption were 0.33, 0.56, and 0.61 mM/h for the 1st, 2nd and 3rd feedings, respectively. This indicated growth of H<sub>2</sub>-consuming organisms.

H<sub>2</sub> consumption could be inhibited by removing CO<sub>2</sub> (Figure 2 – open triangles). After the first H<sub>2</sub> feeding, duplicate vials identical to those above had their headspaces purged with N<sub>2</sub> to remove CO<sub>2</sub> and H<sub>2</sub> and CH<sub>4</sub> (internal standard) were added again. As shown in Figure 2 (open triangles), H<sub>2</sub> was not consumed without CO<sub>2</sub> whereas it was consumed in the vials that still had CO<sub>2</sub> (Figure 2 – closed circles). Park *et al.* (2005) were able to prevent H<sub>2</sub> consumption in batch bottles by removing CO<sub>2</sub> using a KOH trap. This work demonstrates that sludge from a continuous-flow system also consumes H<sub>2</sub> and that, like Park *et al.* (2005), it can be prevented through CO<sub>2</sub> removal.

Additional in-vial activity tests were conducted using several different headspace CO<sub>2</sub> concentrations in the same manner as for Figure 2. The effect of CO<sub>2</sub> concentration on the rate of H<sub>2</sub> consumption is summarised in Figure 3. The rate of H<sub>2</sub> consumption decreased as the CO<sub>2</sub> concentration decreased, with a sharp decline as CO<sub>2</sub> approached zero. Thus, intermediate headspace CO<sub>2</sub> concentrations (1–6 mM) decreased the rate of H<sub>2</sub> consumption without



**Figure 2** | H<sub>2</sub> consumption by sludge from a sparged biohydrogen reactor. Vials were identical for the 1st feeding, then either purged with N<sub>2</sub> (CO<sub>2</sub> = 0.3 mM,  $\Delta$ ) or unchanged (CO<sub>2</sub> = 3.8 mM,  $\bullet$ ).

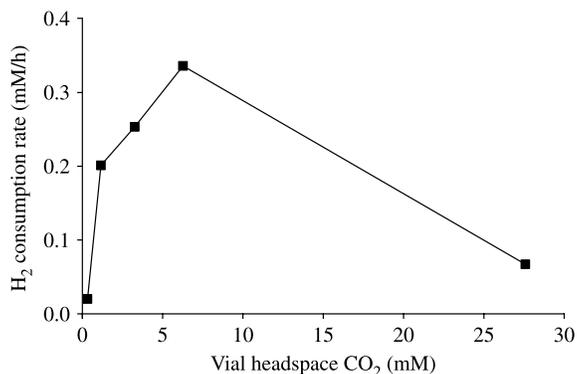
completely inhibiting it. Interestingly, a high CO<sub>2</sub> concentration (27 mM) also decreased H<sub>2</sub> consumption. High CO<sub>2</sub> concentrations are known to be inhibitory (Dixon & Kell 1989) and in biohydrogen systems high CO<sub>2</sub> concentrations in batch bottles have decreased cell growth and H<sub>2</sub> production (Wang *et al.* 2007). While sparging a continuous biohydrogen system with CO<sub>2</sub>, Kim *et al.* (2006) observed a decrease in microbial diversity (including acetogens) and decreased glucose utilization, although the H<sub>2</sub> produced per glucose converted was higher than for N<sub>2</sub> sparging.

A possible mechanism for H<sub>2</sub> consumption in the absence of methanogenesis is homoacetogenesis:

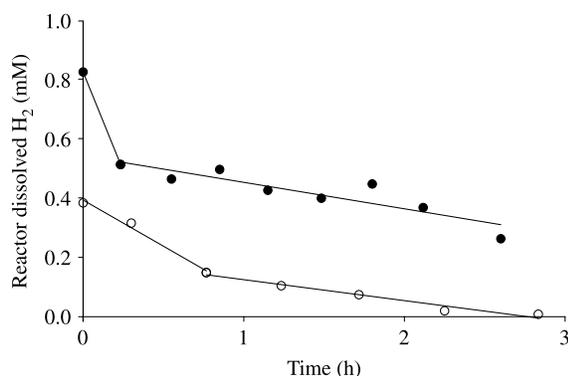


In this study, changes in the acetate concentration during activity tests were undetectable because of the high background acetate from the reactor (Table 1). For example, the acetate concentration in an activity vial could increase by up to 60 mg/L, which is only a 5% change relative to the initial acetate concentration and the same order of magnitude as the reactor variability (Table 1). Park *et al.* (2005) were able to detect changes in acetate because their acetate concentrations were lower (425 mg/L) and expected a larger acetate increase of 235 mg/L.

Figure 4 shows the H<sub>2</sub> consumption tests conducted directly in the reactor. Dissolved H<sub>2</sub> started to decrease immediately after the liquid flows and N<sub>2</sub> sparging were stopped. There were two rates of H<sub>2</sub> removal: a higher initial rate and a slower secondary rate. The initial rates were 0.3 and 1.3 mM/h for sparged and unsparged sludge,



**Figure 3** | In-vial H<sub>2</sub> consumption rate as a function of vial headspace CO<sub>2</sub> concentration. H<sub>2</sub> was consumed by sludge taken from an unsparged continuous-flow biohydrogen reactor.



**Figure 4** | In-reactor dissolved H<sub>2</sub> removal in a fermentative H<sub>2</sub>-producing sludge immediately after stopping all liquid flows and sparging (if in use). Reactor with sparging (○) and without (●).

respectively, although there were not enough data for a statistical analysis. The initial rate included both H<sub>2</sub> consumption as well as H<sub>2</sub> mass-transfer out of solution: initially, the liquid was supersaturated with respect to the headspace, and during this initial period the reactor headspace H<sub>2</sub> concentration increased (not shown). During the second period of H<sub>2</sub> removal, the rates were  $0.071 \pm 0.006$  and  $0.09 \pm 0.01$  mM/h for sparged and unsparged sludge, respectively. Thus, for both periods of H<sub>2</sub> removal sparging may have decreased H<sub>2</sub> consumption.

However, the overall impact of H<sub>2</sub> consumption appears to be small. The H<sub>2</sub> production rates with and without sparging were 10.4 and 7.3 mM/h, respectively, so H<sub>2</sub> consumption for the secondary in-reactor rate was <2% of the H<sub>2</sub> production rate regardless of sparging. And the highest in-vial H<sub>2</sub> consumption rate was 0.34 mM/h (Figure 3), which was <5% of the H<sub>2</sub> production rate. Assuming sparging completely inhibited H<sub>2</sub> consumption, it could account for an H<sub>2</sub> yield of 0.01 and 0.06 mol H<sub>2</sub>/mol glucose converted based on the in-reactor and in-vial consumption rates, respectively, whereas the H<sub>2</sub> yield actually increased by 0.56 mol H<sub>2</sub>/mol glucose converted (Table 1). Consequently, depending on which rate of H<sub>2</sub> consumption is used, preventing H<sub>2</sub> consumption could account for at most 2–11% of the observed H<sub>2</sub> yield increase.

## CONCLUSIONS

This work demonstrated that H<sub>2</sub> consumption exists in an established continuous-flow fermentative H<sub>2</sub> production

system. H<sub>2</sub> consumption decreased as CO<sub>2</sub> decreased, and was inhibited at both low and high CO<sub>2</sub> concentrations. N<sub>2</sub> sparging may have decreased the in-reactor rate of H<sub>2</sub> consumption, although even if sparging did inhibit the highest observed rates of H<sub>2</sub> consumption, it would account for <11% of the observed increase in the H<sub>2</sub> yield during sparging. Therefore, H<sub>2</sub> consumption in continuous fermentative hydrogen production may be of minor concern.

## REFERENCES

- Cruwys, J. A., Dinsdale, R. M., Hawkes, F. R. & Hawkes, D. L. 2002 Development of a static headspace gas chromatographic procedure for the routine analysis of VFAs in wastewaters. *J. Chromatogr. A* **945**(1–2), 195–209.
- Dixon, N. M. & Kell, D. B. 1989 The inhibition by CO<sub>2</sub> of the growth and metabolism of microorganisms. *J. Appl. Bacteriol.* **67**(2), 109–136.
- Hawkes, F. R., Hussy, I., Kyazze, G., Dinsdale, R. & Hawkes, D. L. 2007 Continuous dark fermentative hydrogen production by mesophilic microflora: Principles and progress. *Int. J. Hydrogen Energy* **32**(2), 172–184.
- Hussy, I., Hawkes, F. R., Dinsdale, R. & Hawkes, D. L. 2003 Continuous fermentative hydrogen production from a wheat starch co-product by mixed microflora. *Biotechnol. Bioeng.* **84**(6), 619–626.
- Kim, D. H., Han, S. K., Kim, S. H. & Shin, H. S. 2006 Effect of gas sparging on continuous fermentative hydrogen production. *Int. J. Hydrogen Energy* **31**(15), 2158–2169.
- Kraemer, J. T. & Bagley, D. M. 2006 Supersaturation of dissolved H<sub>2</sub> and CO<sub>2</sub> during fermentative hydrogen production with N<sub>2</sub> sparging. *Biotechnol. Lett.* **28**(18), 1485–1491.
- Kraemer, J. T. & Bagley, D. M. 2007 Improving the yield from fermentative hydrogen production. *Biotechnol. Lett.* **29**(5), 685–695.
- Oh, S. E., Van Ginkel, S. & Logan, B. E. 2005 The relative effectiveness of pH control and heat treatment for enhancing biohydrogen gas production. *Environ. Sci. Technol.* **37**(22), 5186–5190.
- Park, W., Hyun, S. H., Oh, S. E., Logan, B. E. & Kim, I. S. 2005 Removal of headspace CO<sub>2</sub> increases biological hydrogen production. *Environ. Sci. Technol.* **39**(12), 4416–4420.
- Standard Methods for the Examination of Water and Wastewater* 1998 20th edn. American Public Health Association/American Water Works Association/Water Environment Federation, Washington DC, USA.
- Wang, X. J., Ren, N. Q., Xiang, W. S. & Guo, W. Q. 2007 Influence of gaseous end-products inhibition and nutrient limitations on the growth and hydrogen production by hydrogen-producing fermentative bacterial B49. *Int. J. Hydrogen Energy* **32**(6), 748–754.